NanoTBTech



Nanoparticles-based 2D thermal bioimaging technologies H2020-FETOPEN-1-2016-2017 Grant Agreement: 801305



Standardization of light-to-heat conversion efficiency of colloidal nanoheaters

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Standardization of light-to-heat conversion efficiency - presentation plan

- 1. Aim of studies
- 2. Examples of nanoheaters
- 3. Methods of η determining
- 4. Need of standardization
- 5. Results
- 6. A receipt





Why to measure light-toheat conversion efficiency of nanoparticles?

Motivation of study the light-to-heat conversion efficiency - hyperthermia





Motivation of study the light-to-heat conversion efficiency - hyperthermia

Hippocrates: "What medicines to not heal, the lance will; what the lance does not heal, fire will"

How to make hyperthermia more localized?



Whole body hyperthermia







Nanoparticles for hyperthermia treatment - requirements

- Nontoxicity, biocompatibility
- Chemical and photostability
- Nanoparticle size fitting in circulatory/vascular system
- Irradiation wavelength fit in the biological window
- High absorption and light-to-heat conversion efficiency
- Multifunctionality





Hyperthermia nanomaterials



L. Marciniak, K. Kniec, K. Elzbieciak, A. Bednarkiewicz *Non-plasmonic NIR-Activated Photothermal Agents for Photothermal Therapy*. In: Benayas A., Hemmer E., Hong G., Jaque D. (eds) *Near Infrared-Emitting Nanoparticles for Biomedical Applications*. Springer

Other examples of light-to-heat conversion application

• Solar thermal energy

Z. Wang *et al.*, "Dynamic tuning of optical absorbers for accelerated solarthermal energy storage," *Nat. Commun.*, vol. 8, no. 1, 2017.

 Steam generation after solar illumination – possibility of sterillisation

O. Neumann, A. S. Urban, J. Day, S. Lal, P. Nordlander, and N. J. Halas, "Solar vapor generation enabled by nanoparticles," *ACS Nano*, vol. 7, no. 1, pp. 42–49, 2013.



Photoactuators

Y. Yamamoto, K. Kanao, T. Arie, S. Akita, and K. Takei, "Air Ambient-Operated pNIPAM-Based Flexible Actuators Stimulated by Human Body Temperature and Sunlight," ACS Appl. Mater. Interfaces, vol. 7, no. 20, pp. 11002–11006, 2015





How to measure the light-to-heat conversion efficiency - methodology



R. Li, L. Zhang, L. Shi, and P. Wang, "MXene Ti3C2: An Effective 2D Light-to-Heat Conversion Material," ACS Nano, vol. 11, no. 4, pp. 3752–3759, 2017

Light-to-heat conversion efficiency of nanomaterials - models

Roper's model

Temperature probing? mass of all system components



D. K. Roper, W. Ahn, and M. Hoepfner, "Microscale heat transfer transduced by surface plasmon resonant gold nanoparticles," J. Phys. Chem. C, vol. 111, no. 9, pp. 3636–3641, 2007. X. Wang, G. Li, Y. Ding, and S. Sun, "Understanding the photothermal effect of gold nanostars and nanorods for biomedical applications," RSC Adv., vol. 4, no. 57, pp. 30375–30383, 2014

Light-to-heat conversion efficiency of nanomaterials – Chen's model

$$Q_{\text{laser}} = I (1 - \xi) (1 - 10^{-E_{\lambda}}) \eta + I \xi \quad Q_{\text{loss}} = B \Delta T + C (\Delta T)^2$$

Energy balance

$$\left(\mathbf{m}_{\mathrm{s}} c_{\mathrm{p,s}} + \mathbf{m}_{\mathrm{c}} c_{\mathrm{p,c}}\right) \frac{\mathrm{d}\Delta T}{\mathrm{dt}} = I(1-\xi) \left(1-10^{-\mathrm{E}_{\lambda}}\right) \eta + I \xi - B \Delta T - C \left(\Delta T\right)^{2}$$

$$fficiency = \frac{B(T_{end} - T_0) + C(T_{end} - T)^2 - I\xi}{I(1 - \xi)(1 - 10^{-E_{\lambda}})}$$

I – power of the incident laser beam **ξ** - the fraction of the laser energy absorbed by the cuvette walls and the solution ms, cs - mass, heat capacity of the solution mc cc mass and heat capacity of cuvette $\Delta \mathbf{T}$ difference between solution temperature at the time **t** ans starting solution temperature T_0 E_{λ} - the extinction value at the illumination laser wavelength **n** - the ratio of the absorption to extinction

• H. Chen *et al.*, "Understanding the photothermal conversion efficiency of gold nanocrystals," *Small*, vol. 6, no. 20, pp. 2272–2280, 2010.

- Experimental cell in a vacuum Roper
- Cuvette with/without stirrer Chen
- Eppendorf
- Droplet Richardson





- Experimental cell in a vacuum Roper
- Cuvette with a stirrer Chen
- Eppendorf Chernov
- Droplet Richardson
 - + measurements in vacuum
 - + sample cell is tight
 - require dedicated sample cell
 - thermocouple does not touch sample directly



D. K. Roper, W. Ahn, and M. Hoepfner, "Microscale heat transfer transduced by surface plasmon resonant gold nanoparticles," J. Phys. Chem. C, vol. 111, no. 9, pp. 3636–3641, 2007.

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H. Chen *et al.,* "Understanding the photothermal conversion efficiency of gold nanocrystals," *Small*, vol. 6, no. 15 20, pp. 2272–2280, 2010.

- Experimental cell in a vacuum Roper
- Cuvette with a stirrer Chen
- Eppendorf Chernov
- Droplet Richardson
 - + small amount of sample
 required
 + direct observation of
 sample temperature
 region of interest for
 temperature probing not
 specified



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- Experimental cell in a vacuum Roper
- Cuvette with a stirrer Chen
- Eppendorf Chernov
- Droplet Richardson

+small amount of sample required +direct observation of sample temperature -positioning of the droplet against thermocouple/laser beam -evaporation of the sample



H. H. Richardson, M. T. Carlson, P. J. Tandler, P. Hernandez, and A. O. Govorov, "Experimental and theoretical studies of light-to-heat conversion and collective heating effects in metal nanoparticle solutions," *Nano Lett.*, vol. 9, no. 3, pp. 1139–1146, 2009.

Comparison of η for AuNPs

Material (Gold NPs)	Size [nm] (diameter)	Irradiation wavelength [nm]	LSPR wavelength [nm]	Part of mass of a cuvette	Experimental setup comments	Model	ղ _զ [%]	Reference (next slide)
nanocrystals	20	514	524 (broad: 475- 575)	whole glass cell	small sample cell in a vacuum chamber	R*	3.4-9.9	[1]
Au@SiO ₂ nanoshells	154	815	~815	all elements included	vacuum chamber, magnetic stirrer	R	30	[2]
nanorods	44x13	815	780	all elements included	vacuum chamber, magnetic stirrer	R	55	[2]
nanocrystals	20	532	not mentioned?	no cuvette	sample is a droplet	R*	100	[3]
nanocrystals	16-100	809	809	whole cuvette	magnetic stirrer	C*	95-51%	[4]
nanospheres	5-50	532	not mentioned	solvent only	magnetic stirrer, open cuvette	R/C	80-65	[5]
nanostars	~20-120	785	684, 774, 829	"effective mass"	cuvette covered with foam plastic,	W*	79.5-65.2	[6]
nanorods	63.8x24.5, 70.1x19.0, 56.9x13.9	785	683, 774, 821	"effective mass"	magnetic stirrer	W	94.2-69.7	[6]
nanomatryoshkas	100	810	783	whole cuvette		R	63	[7]
nanoshells	150	810	796	whole cuvette		R	39	[7]
nanoshells	15-100	532	520-565?	"offective mass"		W	100-55	[0]
nanorods	10x30-60	700-850	700-850?			W	~100	٥١
nanorods@polymers: chitosan, alginate and poly(vinyl alcohol)	100x50, 150x100, 57x15	808	798, 807 (chitosan)	whole cuvette		R	47–51	[9]
spiky gold	54 - core 58 - spike	808	broad NIR	solvent only		R	78.8	[10]
Bumpy nanospheres	70-90	790	broad (NIR I)	The integral of the temperature		С	96-97	
Smooth nanospheres	56	790	790	gradient was used to determine the effective mass of the cuvette		с	99	[11]
Nanospheres	20	808	524	solvent only	magnetic stirrer	R/C	7.7	[12]
Nanourchins	80		620				25.3	
Nanorods	25x60		650				17.8	
Nanorods	10x41		808				81	
Nanoconjugates (IR806 dye)	20 (Au) 250 (conjugate)		788				31.4	

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Standardization – why results obtained by different labs may differ?

- Efficiency depends on a wavelength
- Differences in existing models
- Effective mass of the sample and holder
- Stirring?
- Region of interest for temperature probing?





The same batch of Au@SiO₂ NPs were applied...



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...and tested in different experimental conditions



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Experimental setup comparison





Light-to-heat conversion efficiency

Material	Size [nm] (diameter)	λ _{εxc} [nm]	Part of mass of a cuvette	Temperature detector	Experimental setup comments	Model	ղ _զ [%]
Au NPs [1]	20	514	whole glass cell	TC outside the measurement cell	small sample cell in a vacuum chamber	R*	3.4-9.9
Au NPs [2]	20	532	no cuvette	TC inside the droplet	sample is a droplet	R*	100
Au NSp [3]	15	532	solvent only	TC inside the cuvette	magnetic stirrer, open cuvette	R/C	78.4
Au NSp [3]	15		-		theoretical value	_	99.6
Au@SiO ₂ [4]	14 (Au), ~140 (Au@SiO ₂)	, 532	solvent only	Measurement through glass	magnetic stirrer on/off	R	57.5 / 50.8
			solvent only	Measurement of sample surface temperature	magnetic stirrer on/off	R	63.1 / 67.5
			cuvette included	Measurement through glass	magnetic stirrer on/off	R	90.5 / 80.8
			cuvette included	Measurement of sample surface temperature	magnetic stirrer on/off	R	98.4 / 106.7
			"effective mass"	Measurement through glass		W	80.5
				Measurement of sample surface temperature	magnetic stirrer on	W	80.6
			no cuvette	Measurement of sample surface temperature	sample is a droplet	R	66.8
						W	81.1

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4. Standardization of light-to-heat conversion efficiency of colloidal nanoheaters, A. Paściak, A. Pilch-Wróbel, Ł. Marciniak, A. Bednarkiewicz (to be submitted)

Laboratory experience – evolution of experimental systems





- Temperature gradients
- Heat receiver – glass
- Large volume of a sample
- Need for calibration



- No temperature gradients
- Heat receiver
 glass less important
- Large volume of a sample
- Need for calibration

- Natural convection only
- Heat receiver tip less important
- Small volume of a sample
- No need for calibration

Miniaturized system performance



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A receipt for a correct measurement

- Stable colloidal sample required
- \geq Reference measurement for solvent should be performed to evaluate Q₀
- Model it is important to be sure what is the effective mass of the setup
- Impact of environment ambient temperature should not change during measurement, evaporation should be minimized
- Use Wang's model rather than Roper's for calculations it does not require full heating/cooling curves





Acknowledgements





Financial support from NanoTBTech-H2020-FETOPEN (801305) "Nanoparticle-based 2D thermal bioimaging technologies, is greatly acknowledged.



This project has received funding from the European Union's Horizon 2020 FET Open programme under Grant Agreement No. 801305.



